19. Synthesis of 4,6,8-Trisubstituted Methyl Azulene-2-carboxylates

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It is shown that sodium (methoxycarbonyl)cyclopentadienide (1), which is easily accessible from sodium cyclopentadienide and dimethyl carbonate in THF, reacts with 2,4,6-trisubstituted pyrylium tetrafluoroborates 2a-d in boiling MeOH to afford the corresponding methyl azulene-2-carboxylates 4a-d in good yields. The corresponding 1-carboxylates 3 were not found (*cf. Schemes 1* and 2).

Whereas the synthesis of azulenes, substituted at the seven-membered ring, can easily be accomplished by the reaction of 2,4,6-trisubstituted pyrylium salts with sodium cyclopentadienide in THF (cf. [1] [2]) or of N-alkylpyridinium salts with sodium cyclopentadienide in DMF (cf. [1] [3]), it is much more problematic to react substituted cyclopentadienides under the above mentioned conditions, since, in general, mixtures of the 1- and 2-substituted azulenes are obtained which are difficult to separate (cf. [1] [4]). The reaction of 2,4,6-trimethylpyrylium perchlorate with sodium methylcyclopentadienide under carefully controlled conditions in THF leads to the formation of 2,4,6,8-tetramethylazulene in moderate yields $[2a]^2$).

We were interested in a plain synthesis of 4,6,8-trisubstituted methyl azulene-2-carboxylates and checked the reaction of sodium (methoxycarbonyl)cyclopentadienide (1), which is easily accessible by the reaction of dimethyl carbonate and sodium cyclopentadienide in THF at 70° [5], with 2,4,6-trimethylpyrylium tetrafluoroborate (2a). In THF, we obtained only mixtures of the corresponding 1- and 2-carboxylates 3a and 4a (*Scheme 1*; *cf.* also [6]).

However, when the reaction was run in boiling MeOH, only the 2-carboxylate **4a** was formed in 47% yield. It was deposited as a microcrystalline powder directly from the boiling solution. It was pure and showed no traces of the 1-carboxylate **3a**. The examples in *Scheme 2* show that the described procedure can generally be applied to the synthesis of 4,6,8-trisubstituted azulene-2-carboxylates.

We assume that, in the protic solvent MeOH, 1 reacts in its 6-methoxyfulvene 6-oxide form, *i.e.*, it should attack the corresponding pyrylium salts with its C(3,4)-atoms, favoring the formation of the 2-carboxylates.

The azulene-2-carboxylates can be reduced in one step to the corresponding 2-methylazulenes with sodium dihydridobis(2-methoxyethoxy)aluminate in the presence of AlCl₃ in toluene (cf. Exper. Part).

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²) Our own experience showed us that the corresponding pyrylium tetrafluoroborate always yields mixtures of 1,4,6,8- and 2,4,6,8-tetramethylazulene.



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Experimental Part

General. See [7] [8]. All reactions were performed under N₂. Et₂O and hexane were distilled before use. Cyclopentadiene (from dicyclopentadiene, *Fluka*, *pract*.) was freshly distilled over a *Vigreux* column before use. NaH (60–65% dispersion in oil, *Fluka*) was washed with hexane before use. THF (*Fluka*, *puriss*. *p.a.*), MeOH (*Merck*, z. A.), dimethyl carbonate (*Fluka*, *purum*), and 2,6-dimethyl-4-pyrone (*Fluka*, *purum*) were used without further purification.

1. Formation of the Methyl Azulene-2-carboxylates. 1.1. Methyl 4,6,8-Trimethylazulene-2-carboxylate (4a). NaH (4.5 g, 0.19 mol) was suspended in THF (190 ml). The suspension was cooled to 0° , and cyclopentadiene (39 ml, 0.38 mol) was added dropwise. The mixture turned red. After all cyclopentadiene had been added, stirring was continued for 1 h at 0° and then the soln. warmed up to r.t. At that point, (MeO)₂CO (50 ml, 0.58 mol) was added dropwise under stirring. After 10 min at r.t., the soln. was heated during 4 h under reflux. The solvent was removed *in vacuo* and the ocker-colored residue suspended in MeOH (250 ml). 2,4,6-Trimethylpyrylium tetrafluoroborate (2a; 21.5 g, 0.11 mol) [9] was added with stirring. The mixture turned at once blue. It was heated under reflux during 12 h. The blue microcrystalline precipitate was filtered, washed with H₂O (2 × 200 ml) and cooled MeOH (2 × 100 ml) to yield pure **4a** (12.1 g, 47% with respect to **2a**). M.p. 175.3–176.7° (Et₂O). $R_{\rm f}$ (hexane/Et₂O 1:1): 0.52. UV (hexane): $\lambda_{\rm max}$ 368.0 (3.77), 351.4 (3.72), 339.2 (3.66, sh), 301.7 (4.70), 291.5 (4.70), 249.5 (4.43), 218.4 (4.09); $\lambda_{\rm min}$ 360.7 (3.54), 310.2 (3.43), 297.5 (4.64), 266.8 (3.83), 227.1 (4.02). IR (CHCl₃): 3021w, 2926w, 1700s, 1540m, 1506m, 1435m, 1324m, 1249m, 1130s. ¹H-NMR (300 MHz, CDCl₃): 7.76 (*s*, H–C(1,3)); 7.05 (*s*, H–C(5,7)); 3.95 (*s*, COOMe); 2.86 (*s*, Me–C(4,8)); 2.61 (*s*, Me–C(6)). ¹³C-NMR (75 MHz, CDCl₃): 168.51 (COOMe); 151.95 (C(6)); 151.88 (C(4,8)); 137.70 (C(3a,8a)); 135.22 (C(2)); 129.92 (C(5,7)); 118.99 (C(1,3)); 53.48 (COOMe); 30.84 (Me–C(6)); 26.94 (Me–C(4,8)). EI-MS: 227 (100, M^{++}), 196 (27, $[M - OCH_3]^+$). Anal. calc. for C₁₅H₁₆O₂ (227.27): C 79.27, H 7.10; found: C 79.02, H 7.03.

1.2. Methyl 6-(tert-Butyl)-4,8-dimethylazulene-2-carboxylate (4b). 1.2.1. 4-(tert-Butyl)-2,6-dimethylpyrylium Tetrafluoroborate (2b). It was prepared in analogy to [10] from 2,6-dimethyl-4-pyrone and t-BuMgCl/Et₂O (crystallized from MeOH/EtOH as long white needles in a yield of 35%). M.p. 200.1–200.9° (MeOH/EtOH). ¹H-NMR (300 MHz, (D₆)acetone): 8.27 (s, H–C(3,5)); 3.02 (s, Me–C(2,6)); 1.49 (s, t-Bu).

1.2.2. Formation of **4b**. The ester was prepared in analogy to **4a** starting with NaH (1.2 g, 50 mmol), THF (50 ml), cyclopentadiene (10.3 ml, 100 mmol), (MeO)₂CO (13.2 ml, 153 mmol), **2b** (4.78 g, 29 mmol), and MeOH (65 ml). The ester **4b** (3.0 g; 38 % with respect to **2b**) was obtained as a blue microcrystalline powder. M.p. 160.5–162.1° (hexane/AcOEt). $R_{\rm f}$ (hexane/Et₂O 1:1): 0.61. UV (hexane): $\lambda_{\rm max}$ 366.2 (3.86), 349.5 (3.80), 296.0 (4.80), 285.8 (4.81), 244.5 (4.53), 213.2 (4.16); $\lambda_{\rm min}$ 358.5 (3.65), 339.2 (3.77), 320.8 (3.69), 291.3 (4.77), 258.7 (4.27), 221.0 (4.13). IR (KBr): 2996w, 2961w, 1710s, 1575m, 1323m, 1221s. ¹H-NMR (300 MHz, CDCl₃): 7.76 (s, H–C(1,3)); 7.33 (s, H–C(5,7)); 3.96 (s, COOMe); 2.93 (s, Me–C(4,8)); 1.46 (s, t-Bu). ¹³C-NMR (75 MHz, CDCl₃): 167.17 (COOMe); 162.15 (C(6)); 149.69 (C(4,8)); 135.97 (C(3a,8a)); 133.86 (C(2)); 124.74 (C(5,7)); 116.65 (C(1,3)); 51.57 (COOMe); 38.94 (Me₃C); 31.84 (Me₃C); 25.71 (Me–C(4,8)). MS: 270 (100, M^{+-}), 255 (11, $[M - CH₃]^+$), 239 (9, $[M - OCH₃]^+$). Anal. calc. for C₁₈H₂₂O₂ (270.37): C 79.97, H 7.20; found: C 79.76, H 7.15.

1.3. Methyl 4,8-Dimethyl-6-phenylazulene-2-carboxylate (4c). 1.3.1. 2,6-Dimethyl-4-phenylpyrylium Tetrafluoroborate (2c). It was prepared from 2,6-dimethyl-4-pyrone and PhMgBr/Et₂O (crystallized from MeOH/EtOH as red-brown platelets and needles in a yield of 73%). M.p. 201.9-202.4° (MeOH/EtOH). ¹H-NMR (300 MHz, (D₆)acetone): 8.61 (s, H-C(3,5)); 8.33 (d (f.s.), ³J = 7.2, H-C(2,6) of Ph); 7.88 (t (f.s.), ³J = 7.2, H-C(4) of Ph); 7.77 (t (f.s.), ³J = 7.2, H-C(3,5) of Ph); 3.08 (s, Me-C(2,6)).

1.3.2. Formation of 4c. The ester was prepared in analogy to 4a starting with NaH (1.2 g, 50 mmol), THF (50 ml), cyclopentadiene (10.3 ml, 100 mmol), (MeO)₂CO (13.2 ml, 153 mmol), 2c (5.37 g, 29 mmol), and MeOH (65 ml). The ester 4c (3.80 g; 45% with respect to 2c) was obtained as a blue-greenish microcrystalline powder. M.p. 165.3–167.5° (hexane/AcOEI). R_f (hexane/Et₂O 1:1): 0.49 UV (hexane): λ_{max} 377.4 (4.02), 346.0 (3.74, sh), 333.5 (3.66), 304.1 (4.75), 295.7 (4.69, sh), 248.5 (4.26), 212.9 (3.93); λ_{min} 338.2 (3.63), 320.8 (3.63), 262.0 (3.56), 224.6 (3.83). IR (KBr): 3022w, 2971w, 1714s, 1577m, 1452m, 1329m, 1230s. ¹H-NMR (300 MHz, CDCl₃): 7.86 (s, H–C(1,3)); 7.63 (dd, ³J = 6.8, ⁴J = 1.3, H–C(2,6) of Ph); 7.52–7.43 (m, H–C(3,4,5) of Ph); 7.37 (s, H–C(5,7)); 3.98 (s, COOMe); 2.97 (s, Me–C(4.8)). ¹³C-NMR (75 MHz, CDCl₃): 165.07 COOMe); 150.14 (C(4,8)); 145.27 (C(6)); 136.11 (C(3a,8a)); 134.32 (C(2)); 134.03 (C(1) of Ph); 128.54 (2 arom. C); 128.54 (2 arom. C); 128.16 (C(4) of Ph); 127.31 (C(5,7)); 177.44 (C(1,3)); 51.93 (COOMe); 25.39 (Me–C(4.8)). MS: 290 (100, M^+), 259 (49, $[M - OCH_3]^+$). Anal. calc. for $C_{20}H_{18}O_2$ (290.36): C 82.74, H 6.25; found: C 82.73, H 6.21.

1.4. Methyl 6-(4-Methoxyphenyl)-4,8-dimethylazulene-2-carboxylate (4d). 1.4.1. 4-(4-Methoxyphenyl)-2,6-dimethylpyrylium Tetrafluoroborate (2d). It was prepared from 2,6-dimethyl-4-pyrone and (4-MeOC₆H₄)MgBr/Et₂O (crystallized from MeOH/EtOH as a microcrystalline yellowish powder). M.p. 203.8-204.6° (MeOH/EtOH). ¹H-NMR (300 MHz, (D₆)acetone): 8.56 (*s*, H-C(3,5)); 8.45 (*d* (f.s.), ³J = 9.0, H-C(2,6) of 4-MeOC₆H₄); 7.33 (*t* (f.s.), ³J = 9.0, H-C(3,5) of 4-MeOC₆H₄); 4.04 (*s*, CH₃O); 2.96 (*s*, Me-C(2,6)).

1.4.2. Formation of **4d**. The ester was prepared in analogy to **4a** starting with NaH (1.2 g, 50 mmol), THF (50 ml), cyclopentadiene (10.3 ml, 100 mmol), (MeO)₂CO (13.2 ml, 153 mmol), **2a** (5.37 g, 29 mmol), and MeOH (65 ml). The ester **4d** (3.90 g; 42% with respect to **2d**) was obtained as a microcrystalline greenish powder. M.p. 169.8-172.5° (hexane/AcOEt). $R_{\rm f}$ (hexane/Et₂O 1:1): 0.45. UV (hexane): $\lambda_{\rm max}$ 381.1 (4.32), 320.5 (4.78), 308.7 (4.71), 292.5 (4.61), 243.0 (4.50, sh), 233.4 (4.52); $\lambda_{\rm min}$ 356.1 (4.12), 312.8 (4.69), 297.0 (4.60), 263.9 (4.26). IR (KBr): 3042w, 2976w, 1718s, 1576m, 1516m, 1441m, 1326m, 1298m, 1264m, 1223s. ¹H-NMR (300 MHz, CDCl₃): 7.83 (s, H-C(1,3)); 7.58 (d, ³J = 8.5, H-C(2,6) of 4-MeOC₆H₄); 7.35 (s, H-C(5,7)); 7.02 (d, ³J = 8.8, H-C(3,5) of 4-MeOC₆H₄); 3.98 (s, COOMe); 3.88 (s, MeO); 2.96 (s, Me-C(4,8)). ¹³C-NMR (75 MHz, CDCl₃): 165.83 (COOMe); 151.71 (C(6)); 150.13 (C(4,8)); 137.68 (C(4) of 4-MeOC₆H₄); 135.94 (C(3a,8a)); 133.67 (C(2)); 129.94 (C(2,6) of 4-MeOC₆H₄); 127.99 (C(5,7)); 117.46 (C(1,3)); 114.25 (C(3,5) of 4-MeOC₆H₄); 55.46 (MeO); 51.75 (COOMe); 25.52 (Me-C(4,8)). MS: 320 (100, M⁺), 289 (31, [M - OCH₃]⁺). Anal. calc. for C₂₁H₂₀O₃ (320.39): C 78.72, H 6.29; found: C 78.62, H 6.29.

2. Reduction of 4a. AlCl₃ (0.53 g, 4.0 mmol; Fluka, puriss.) was added to toluene (10 ml). After cooling to 0°, sodium dihydridobis(2-methoxyethoxy)aluminate in toluene (4.6 ml, 16 mmol; Fluka, pract. 70% in toluene) was added and the mixture stirred for 1 h at 0°. After this time, a soln. of 4a (0.86 g, 4 mmol) in toluene (20 ml) was added dropwise. The temp. went up to 28–30°, and the color of the mixture changed from blue to red-blue. TLC showed that most of 4a had reacted to 2-(hydroxymethyl)-4,6,8-trimethylazulene, and only traces of 2,4,6,8-tertramethylazulene was detectable by TLC. An aq. soln. of NaOH (20 ml, 20%) was added and the colorless precipitate removed by filtration. The filtrate was extracted with Et₂O, and the combined org. layers, after washing with H₂O, were dried (MgSO₄). Evaporation of the solvent yielded pure 2,4,6,8-tetramethylazulene (0.65 g, 90%; cf. [10]). M.p. 99.8-101.1° (EtOH). R_f (hexane/Et₂O 1:1): 0.71. UV (hexane): λ_{max} 351.8 (3.74), 313.3 (3.82), 293.6 (4.76), 285.2 (4.71), 245.6 (4.45); λ_{min} 342.7 (3.60), 311.8 (3.78), 287.3 (4.73), 258.5 (3.63), 220.9 (4.23). IR (KBr): 3005w, 2923w, 1578m, 1501m, 1447m, 1375w, 1333m, 1312w, 1088w, 1027w. ¹H-NMR (300 MHz, CDCl₃): 7.12 (s, H-C(1,3)); 7.02 (s, H-C(5,7)); 2.82 (s, Me-C(4,8)); 2.61, 2.59 (2s, Me-C(2,6)).

REFERENCES

- K.-P. Zeller, in 'Houben-Weyl, Methoden der organischen Chemie', 4. Aufl. Bd. 5/2c, 'Carbocyclische π-Elektronen-Systeme', Georg Thieme Verlag, Stuttgart, New York 1985, 127ff.
- [2] a) K. Hafner, H. Kaiser, Liebigs Ann. Chem. 1958, 618, 140; b) K. Hafner, H. Kaiser, Org. Synth., Coll. Vol. 1973, 5, 1088.
- [3] R.W. Alder, G. Whittaker, J. Chem. Soc., Perkin Trans. 2 1975, 714.
- [4] K. Hafner, Angew. Chem. 1958, 70, 419.
- [5] W.P. Hart, D. Shihua, M.D. Rausch, J. Organomet. Chem. 1985, 282, 111.
- [6] a) A.A.S. Briquet, H.-J. Hansen, Helv. Chim. Acta 1994, 77, 1577. b) A.A.S. Briquet, Ph.D. Thesis, University of Zurich, 1994.
- [7] A.J. Rippert, H.-J. Hansen, Helv. Chim. Acta 1992, 75, 2219.
- [8] A. J. Rippert, H.-J. Hansen, Helv. Chim. Acta 1993, 76, 2906.
- [9] A.T. Balaban, A.J. Boulton, Org. Synth., Coll. Vol. 1973, 5, 1112.
- [10] Y. Chen, R.W. Kunz, P. Uebelhart, R.H. Weber, H.-J. Hansen, Helv. Chim. Acta 1992, 75, 2447; R.H. Weber, Ph. D. Thesis, University of Basel, 1988.